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transformation
experiments. Griffith
took 4 mice and
injected each one of
them with a different

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bacteria. They were the S bacteria, the R bacteria, the Heat-killed S bacteria, and a mixture ...

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Stoichiometry

Complete the following
sentences by 'filling in
the appropriate word
or phrase from the list
below. Each word or
phrase maybe used

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once, more than once,
or not at all reactants
molar ratio quantitative
actual particles.
condraption of matter
coefficients subscripts
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ANSWER KEY 7.

opposes the motion of objects that move through the air, is affected by speed, size, and shape 8. net force

9. microwelds 10.

rolling 11. air

resistance 12.

acceleration 13. sliding

14. parachute 1.

Gravity is a force that every object in the

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11—2 Review and
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Stoichiometry
Problems If the
statement is true, write
"true." If it is false,
change the underlined
word or words to make
it true. Write your
answer on the line
provided. 1. 2. 5. The
major types of
stoichiometry problems
are mass-mass, mass-
volume, and volume-

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volume.

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Below you will find a description of each lesson within each section of Chapter 9. There is a link for the HW Help (this can be found within the HW problems on your Ebook as well).

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equilibrium is established when the Equid vaporizes at the same rate that vapor condenses. 7, NO, in reversible reactions at equilibrium,

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Chemistry 101

ANSWER KEY 1 REVIEW

QUESTIONS Chapter 9

1. Draw Lewis

structures for each of the following structures and assign formal charges to each atom:

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- a) SF₂ 20 electrons 0
0 0 b) NH₂ OH (N and
O are bonded to one
another) 14 electrons 0
0 0 0 0 c) PO₄³⁻

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24—1 Review and
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(continued) 13. There
are 3.29 g of

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iodine-126 remaining in a sample originally containing 26.3 g of iodine-126. The half-life of iodine-126 is 13 days. How old is the sample? 14. A sample that originally contained 2.5 g of rubidium-87 now contains 1.25 g. The half-life of rubidium-87 is 6×10^{10} years.

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2 Answer.Key 2-1

Explore, page 5 The original source of energy for this activity would be the stored chemical energy in the ... 2-1 Review and Reinforcement, pages 8-9 . 1. energy 2. radiant 3. chemical potential 4. joule 5. created, destroyed 6. natural gas 7. kinetic . 8. f . 9. c 10. b 11. a .

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