

Chemthink Molecular Shapes Answers

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Chemthink Molecular Shapes Answers

In this Chemthink tutorial, you will explore molecular shapes and the VSEPR theory and take a short quiz. Topics include: attraction and repulsion between charged particles. VSEPR (Valence Shell Electron Pair Repulsion) Theory. going from the 2-dimensional Lewis Dot representation to a 3-dimensional visualization of the molecule.

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ChemThink: Molecular Shapes VSEPR theory tells us that areas of a molecule that have a (Higher/Lower) concentration of _electrons_ will (Repel/Attract) each other and that these _Repulsions_ will affect the _ Shape_ of the molecule. Areas of e - concentra tion Draw an Example Bond Angles Shape Ex. Lewis Dot (Circle the e - concentration) 2 180 O Linear 3 Trigonal Planar 3 Bent 4 Tetrahedral Valence Shell Electron Pair R _____.

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ChemThink: Molecular Shapes VSEPR theory tells us that areas of a molecule that have a (Higher/Lower) concentration of ____ will (Repel/Attract) each other and that these ____ will affect the ____ of the molecule.

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Molecular Shapes, Molecular Shapes. bent. tetrahedral. linear. trigonal planar. bond angle 104.5 degrees... 3 electron groups ... 2 bonding groups.... bond angles 109.5 degrees... 4 electron groups ... 4/3/2 bonding gro.... bond angle 180 degrees... 2/5 electron groups ... 2 bonding groups... 0....

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Chemthink Ionic Bonding Answers Chemthink Ionic Bonding Answers In this Chemthink tutorial, you will explore ionic bonding and take a short quiz. Topics include: how ions form and why they stick together; the 3-dimensional shape of an ionic crystal; counting ions in a crystal to determine the ionic formula Chemthink - Ionic Page 5/13 Answers To ...

Chemthink Covalent Bonding Answers

Understand why, within a covalently-bonded molecule, areas with a higher concentration of electrons repel each other to be as far apart as possible. See how Lewis structures can be used to predict the shape of a molecule, and learn about common molecular geometries such as linear, trigonal planar, bent, tetrahedral, and trigonal pyramid.

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Chemthink - Molecular Shapes. Student Worksheet. This Chemthink module helps students learn how to construct Lewis dot structures for covalent compounds and predict their molecular shapes with the VSEPR theory. Essential concepts:Covalent bonding, Lewis dot structures, molecular geometry, VSEPR. Answer key:Included in the chemistry instructor resources subscription.

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